

266. *The Preparation of 1 : 8-Naphthyridines from
2 : 6-Diaminopyridine.*

By V. PETROW, E. L. REWALD, and B. STURGEON.

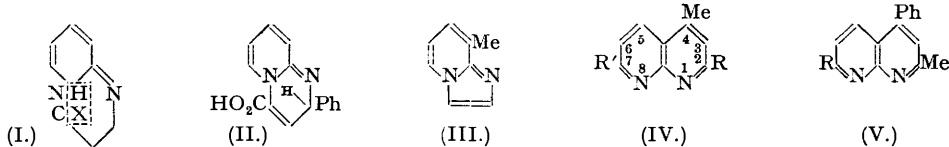
7-Amino-2-hydroxy-4-methyl-1 : 8-naphthyridine (IV; R = OH, R' = NH₂), prepared by condensation of ethyl acetoacetate or ethyl β -amimicrotonate with 2 : 6-diaminopyridine, has been converted into *2-chloro-7-acetamido-4-methyl-1 : 8-naphthyridine* (IV; R = Cl, R' = NHAc), and hence into *2-anilino-*, *2-p-chloroanilino-*, *2-piperidino-*, and *2-phenoxy-7-amino-4-methyl-1 : 8-naphthyridine*.

Condensation of benzoylacetone with 2:6-diaminopyridine gives 7-amino-4-phenyl-2-methyl-1:8-naphthyridine (V; R = NH₂), converted into 7-hydroxy-, 7-chloro-, and hence into 7-anilino-, 7-piperidino-, and 7-phenoxy-4-phenyl-2-methyl-1:8-naphthyridine. (V; R = NHAc) gives the 1(N)-monoquaternary salt.

OUR interest in the naphthyridines arose from the chance observation that certain members of the series possessed valuable biological properties (see Petrow, *J.*, 1946, 200), and a comprehensive survey of the field is now in progress in these Laboratories. The present communication describes some new derivatives of 1:8-naphthyridine. Biological data will be reported elsewhere.

Although 2-aminopyridine would appear to form the obvious starting point for the synthesis of 1:8-naphthyridines, they have not hitherto been prepared from this base. Earlier claims to this effect by Reissert (*Ber.*, 1895, 28, 119) and R ath (*Annalen*, 1931, 486, 284; see also E.P.P. 319,974, 339,932 and G.P.P. 526,630, 522,272) have been conclusively disproved by Bose and Sen (*J.*, 1931, 2840), Seide and Chelinzwen (*J. Gen. Chem. Russia*, 1937, 7, 2314), Sp ath and Kuffner (*Ber.*, 1938, 71, 1657), and Petrow (*J.*, 1945, 928). It would appear that 2-aminopyridine reacts preferentially in the imino-form [partial formula (I)] to give the pyrimidine ring system such as (II) in cases of ring closure. Reactions with ethyl acetoacetate (Crippa and Scevola, *Gazzetta*, 1937, 67, 330), ethyl benzoylacetate (Seide, *Ber.*, 1925, 58, 352; cf. Palazzo and Tamburini, *Atti R. Accad. Lincei*, 1911, 20, I, 37), ethyl malonate (Tschitschibabin, *Ber.*, 1924, 57, 1169), and ethylene oxide (Knunjanz, *Ber.*, 1935, 68, 397) have been shown to fall into this category. The recent claim of Mazza and Migliardi (*Atti R. Accad. Sci. Torino*, 1940, 75, I, 438; see also Migliardi, *ibid.*, p. 548) that reaction of 2-aminopyridine with benzaldehyde and pyruvic acid leads to a 4-carboxy-2-phenyl-1:8-naphthyridine cannot be accepted. There seems little doubt that their product should be correctly formulated as the pyrimidine derivative (II). Rath (*Ber.*, 1925, 58, 347; cf. Tschitschibabin, *ibid.*, p. 1704) has described the formation of a "1:2-dihydro-1:8-naphthyridine" in low yield from the products of reaction of 2-amino-3-methylpyridine with bromoacetal at 250°. This claim, too, must be accepted with caution, as it seems unreasonable to assume the stability of a 1:2-dihydro-structure under the conditions of the experiment. In our view, this reaction leads to the 8-methylpyrimidazole (III). The facile formation of ring systems of this type has been demonstrated by Schmidt and Bangler (*Ber.*, 1926, 59, 1360), Tschitschibabin (*loc. cit.*), and Allen *et al.* (*J. Amer. Chem. Soc.*, 1944, 66, 1805). 2-Aminopyridine behaves as a cyclic amidine in these reactions, and on electro-chemical grounds alone its conversion into a 1:8-naphthyridine appears highly improbable.

2:6-Diaminopyridine, in striking contrast to 2-aminopyridine, forms the most accessible starting material for the synthesis of 1:8-naphthyridines, owing probably to the presence of two amino-groupings in the molecule either of which can take part in the prototropic change involved in imine-formation. By direct condensation of the base with ethyl acetoacetate, Seide (*Ber.*, 1926, 59, 2465) obtained 7-amino-2-hydroxy-4-methyl-1:8-naphthyridine (IV; R = OH, R' = NH₂) in good yield. The reaction has been extended to ethyl benzoylacetate (Mangini and Colonna, *Gazzetta*, 1942, 72, 183), acetylacetone (Mangini, *Chem. Zentr.*, 1940, II, 2613; Ochiai and Miyaki, *Ber.*, 1941, 74, 1115), and ethoxymethylene diethylmalonate (Adams *et al.*, *J. Amer. Chem. Soc.*, 1946, 68, 1317).



We now find that 7-amino-2-hydroxy-4-methyl-1:8-naphthyridine (IV; R = OH, R' = NH₂) (Seide, *loc. cit.*) passes smoothly under the action of acetic anhydride into 7-acetamido-2-hydroxy-4-methyl-1:8-naphthyridine (Mangini and Colonna, *loc. cit.*). The acetamido-grouping in this compound proved stable to phosphorus oxychloride, giving 2-chloro-7-acetamido-4-methyl-1:8-naphthyridine (IV; R = Cl, R' = NHAc), occasionally mixed with some 2-chloro-7-amino-4-methyl-1:8-naphthyridine. Treatment of (IV; R = Cl, R' = NHAc) with phenol at 180° gave 7-acetamido-2-phenoxy-4-methyl-1:8-naphthyridine, hydrolysed to the corresponding amino-derivative. 7-Amino-2-anilino-, 7-amino-2-p-chloroanilino-, and 7-amino-2-piperidino-4-methyl-1:8-naphthyridine were similarly prepared.

Condensation of 2:6-diaminopyridine with benzoylacetone in the presence of zinc chloride

gave 7-amino-4-phenyl-2-methyl-1 : 8-naphthyridine (V; R = NH₂), characterised by preparation of the *picrate*. The derived *acetamido*-compound (V; R = NHAc) was converted *via* the methosulphate into 7-acetamido-4-phenyl-2-methyl-1 : 8-naphthyridine-1-methiodide, and hence into the corresponding *methochloride*. The constitution assigned to the monomethiodide followed from the formation of a highly coloured dye on heating the compound with *p*-dimethylaminobenzaldehyde in alcoholic solution containing a trace of piperidine. This colour change is almost certainly due to the formation of the corresponding 2-*p*-dimethylamino-styryl derivative, but attempts to isolate it in a state of analytical purity were unsuccessful. Treatment of (V; R = NH₂) with nitrous acid gave 7-hydroxy-4-phenyl-2-methyl-1 : 8-naphthyridine (V; R = OH), from which 7-chloro-, and hence 7-phenoxy-, 7-anilino-, and 7-piperidino-4-phenyl-2-methyl-1 : 8-naphthyridine were readily prepared. Attempts to condense (V; R = NH₂) with *p*-acetamidobenzenesulphonyl chloride were unsuccessful.

We have *inter alia* examined the condensation of ethyl β-aminocrotonate with 2 : 6-diaminopyridine in the hope of obtaining the isomeric 7-amino-4-hydroxy-2-methyl-1 : 8-naphthyridine, but the only product isolated as (IV; R = OH, R' = NH₂) in somewhat better yield. The corresponding nitrile, β-aminocrotononitrile, it may be added, reacts with arylamines with exceptional facility to give the β-aryliminobutyronitriles by loss of the elements of ammonia (Meyer, *J. pr. Chem.*, 1908, **78**, 499). Its behaviour with the aminopyridines will form the subject of a later communication.

EXPERIMENTAL.

(M. p.s are corrected. Microanalyses are by Drs. Weiler and Strauss, Oxford.)

2-Chloro-7-acetamido-4-methyl-1 : 8-naphthyridine (IV; R = Cl, R' = NHAc).—7-Acetamido-2-hydroxy-4-methyl-1 : 8-naphthyridine was prepared by heating finely powdered (IV; R = OH, R' = NH₂) (Seide, *Ber.*, 1926, **59**, 2465) (2 g.) with acetic anhydride (30 ml.) under reflux for 30 mins. The heavy yellow solid changed to a flocculent white suspension. When cold, the product was collected and recrystallised from a large volume of glacial acetic acid (charcoal), giving white crystals, m. p. > 310° (Found: C, 60.8; H, 5.2; N, 19.0. Calc. for C₁₁H₁₁O₂N₃ C, 60.8; H, 5.1; N, 19.3%). Yield 2.2 g. (89%) (Mangini and Colonna, *loc. cit.*, give m. p. < 285°). The acetamido-compound (10 g.) was heated with phosphorus oxychloride (100 ml.) for 30 minutes under reflux. The product was decomposed with ice-water and made alkaline with sodium hydroxide, and the precipitated solids were recrystallised from spirit. 2-Chloro-7-acetamido-4-methyl-1 : 8-naphthyridine formed felted white needles (7.5 g.; 69%), m. p. 240° (Found: C, 55.9; H, 3.8; N, 17.8; Cl, 15.6. C₁₁H₁₀ON₃Cl requires C, 56.1; H, 4.2; N, 17.8; Cl, 15.1%).

If the temperature rose somewhat during the decomposition of the phosphorus oxychloride reaction mixture, 2-chloro-7-amino-4-methyl-1 : 8-naphthyridine, white needles, m. p. 258—259°, from alcohol (Found: Cl, 21.3. C₉H₈N₃Cl requires Cl, 21.7%), was also obtained, readily separated from the acetamido-compound by fractionation from alcohol in which it was less soluble. On acetylation it gave the acetamido-derivative, m. p. 240° (above).

7-Acetamido-2-phenoxy-4-methyl-1 : 8-naphthyridine.—2-Chloro-7-acetamido-4-methyl-1 : 8-naphthyridine (1 g.) was heated with phenol (3 g.) at 180° for 1 hour. The cooled melt was treated with excess of sodium hydroxide solution, and the insoluble fraction collected and heated with acetic anhydride on the water-bath. 7-Acetamido-2-phenoxy-4-methyl-1 : 8-naphthyridine, separated on cooling as long needles from aqueous alcohol, m. p. 205° (Found: C, 69.3; H, 5.2; N, 14.1. C₁₇H₁₅O₂N₃ requires C, 69.6; H, 5.1; N, 14.3%). Hydrolysis with alcoholic hydrochloric acid gave 7-amino-2-phenoxy-4-methyl-1 : 8-naphthyridine, needles from benzene, m. p. 216—217° (Found: C, 71.6; H, 5.3; N, 16.7. C₁₅H₁₃ON₃ requires C, 71.8; H, 5.2; N, 16.7%).

7-Amino-2-anilino-4-methyl-1 : 8-naphthyridine (IV; R = NHPh, R' = NH₂).—2-Chloro-7-acetamido-4-methyl-1 : 8-naphthyridine (4 g.) was heated with aniline (10 ml.) under reflux for 3 hours. Excess of aniline was removed in steam, and the solids were collected and heated under reflux with 2N-hydrochloric acid for 1 hour. The mixture was made alkaline with sodium hydroxide and the product purified from spirit. 7-Amino-2-anilino-4-methyl-1 : 8-naphthyridine formed yellow crystals, m. p. 269—270° (Found: C, 71.6; H, 5.4; N, 23.0. C₁₅H₁₄N₄ requires C, 72.0; H, 5.6; N, 22.4%).

7-Amino-2-*p*-chloroanilino-4-methyl-1 : 8-naphthyridine.—2-Chloro-7-acetamido-4-methyl-1 : 8-naphthyridine (2.5 g.) and *p*-chloroaniline (2.5 g.) were heated under reflux in nitrobenzene (25 ml.). Vigorous reaction occurred in a few minutes and the mixture set to a semi-solid. Heating at 200° was continued for one hour. The yellow-brown solids were collected and crystallised from alcohol-light petroleum. 7-Amino-2-*p*-chloroanilino-4-methyl-1 : 8-naphthyridine formed ivory needles (1 g.; 33%), m. p. 224—225° (Found: C, 63.1; H, 4.9; N, 19.5; Cl, 12.5. C₁₅H₁₃N₄Cl requires C, 63.3; H, 4.6; N, 19.7; Cl, 12.5%). The *hydrochloride* formed yellow needles from alcohol-light petroleum, m. p. > 300° (Found: Cl, 22.8. C₁₅H₁₄N₄Cl₂ requires Cl, 22.9%).

7-Amino-2-piperidino-4-methyl-1 : 8-naphthyridine.—(IV; R = Cl, R' = NHAc) (1 g.) was heated with piperidine (5 ml.) at 180° for 8 hours in a sealed tube. Excess of piperidine was removed on the water-bath. The crystalline product, after hydrolysis with concentrated hydrochloric acid (10 ml.) and alcohol (25 ml.) for 1 hour, followed by basification with sodium hydroxide, gave 7-amino-2-piperidino-4-methyl-1 : 8-naphthyridine, ivory crystals from light petroleum, m. p. 221.5—222.5° (Found: C, 69.0; H, 7.2; N, 22.7. C₁₄H₁₈N₄ requires C, 69.4; H, 7.4; N, 23.1%).

7-Amino-4-phenyl-2-methyl-1 : 8-naphthyridine (V; R = NH₂).—2 : 6-Diaminopyridine (20 g.), benzoylacetone (30 g.), and powdered anhydrous zinc chloride (12 g.) were heated in an oil-bath. Reaction

commenced at 135°, after which the temperature was slowly raised to 175° over 3½ hours. Solution in alcohol (200 ml.) followed by gradual precipitation with water (300 ml.) gave a granular product, further purified by solution in 2*N*-hydrochloric acid (charcoal) and precipitation with sodium hydroxide. For purification, the crude naphthyridine (8 g.) and acetic anhydride (80 ml.) were heated under reflux for 30 minutes. 7-Acetamido-4-phenyl-2-methyl-1:8-naphthyridine, needles from benzene (30%), m. p. 207—208° (Found: C, 73.4; H, 5.4; N, 15.0. C₁₇H₁₅ON₃ requires C, 73.6; H, 5.4; N, 15.2%), separated on cooling. Hydrolysis with boiling 15% hydrochloric acid for 30 minutes gave 7-amino-4-phenyl-2-methyl-1:8-naphthyridine, pale yellow octahedra from nitrobenzene, m. p. 247.5—248.5° (Found: C, 76.4; H, 5.6; N, 18.1. C₁₅H₁₃N₃ requires C, 76.6; H, 5.5; N, 17.9%), characterised as the *picrate*, fine yellow needles from a large volume of alcohol, m. p. 284° (decomp.) (Found: N, 18.1. C₁₅H₁₃N₃, C₆H₃O₇N₃ requires N, 18.1%).

7-Acetamido-4-phenyl-2-methyl-1:8-naphthyridine-1-(*N*)-methiodide, prepared *via* the methosulphate, formed silky yellow needles (75%) from alcohol, m. p. 244—245° (Found: I, 31.2. C₁₈H₁₈ON₂I requires I, 30.4%). The methochloride formed pale green platelets from alcohol-ether, m. p. 235—236° (Found: Cl, 11.1. C₁₅H₁₃ON₂Cl requires Cl, 10.8%).

7-Hydroxy-4-phenyl-2-methyl-1:8-naphthyridine (V; R = OH).—The amino-compound (V; R = NH₂) (10 g.) in water (250 ml.) and concentrated hydrochloric acid (5 ml.), mechanically stirred at 0°, was treated simultaneously during 30 minutes with sodium nitrite (4.5 g. in 20 ml. water) and hydrochloric acid (10 ml. in 30 ml. water). After 2 hours the mixture was heated on the water-bath until evolution of nitrogen had ceased. The product was precipitated with ammonium hydroxide (*d* 0.88), giving 7-hydroxy-4-phenyl-2-methyl-1:8-naphthyridine, needles from alcohol, m. p. 252—253° (Found: C, 76.3; H, 5.2; N, 11.6. C₁₅H₁₂ON₂ requires C, 76.2; H, 5.1; N, 11.9%). The compound dissolves in sodium hydroxide solution.

7-Chloro-4-phenyl-2-methyl-1:8-naphthyridine (V; R = Cl).—The hydroxy-compound (V; R = OH) (2 g.) and phosphorus oxychloride (15 ml.) were heated under reflux at 140—150° for 30 minutes. The product, isolated in the usual way, gave 7-chloro-4-phenyl-2-methyl-1:8-naphthyridine (70%), felted needles from aqueous methanol (charcoal), m. p. 161° (Found: Cl, 14.4. C₁₅H₁₁N₂Cl requires Cl, 14.0%).

7-Phenoxy-4-phenyl-2-methyl-1:8-naphthyridine separated from benzene-light petroleum in needles (80%), m. p. 156—157.5° (Found: C, 80.6; H, 5.3; N, 8.9. C₂₁H₁₆ON₂ requires C, 80.7; H, 5.2; N, 8.9%).

7-Anilino-4-phenyl-2-methyl-1:8-naphthyridine formed pale yellow needles from a large volume of alcohol, m. p. 286—287° (Found: C, 80.4; H, 5.6; N, 13.3. C₂₁H₁₇N₃ requires C, 81.0; H, 5.5; N, 13.5%).

7-Piperidino-4-phenyl-2-methyl-1:8-naphthyridine, isolated as the *picrate*, long yellow needles from alcohol, m. p. 220—221° (decomp. at 238°) (Found: N, 15.7. C₂₀H₂₁N₃, C₆H₃O₇N₃ requires N, 15.8%), formed pale yellow plates from benzene-light petroleum, m. p. 131—132° (Found: C, 79.0; H, 7.0; N, 14.0. C₂₀H₂₁N₃ requires C, 79.2; H, 6.9; N, 13.9%).

7-Amino-2-hydroxy-4-methyl-1:8-naphthyridine (IV; R = OH, R' = NH₂).—2:6-Diaminopyridine (5 g.) and ethyl β-aminocrotonate (6 g.) were heated at 180—200° for 2 hours, after which the temperature was raised to 220° during 45 minutes, the mixture then solidifying. The grey-green mass was extracted with alcohol, giving (IV; R = OH, R' = NH₂), m. p. >360°, identified by acetylation and chlorination to (IV; R = Cl, R' = NHAc), m. p. 240°, alone or in admixture with an authentic specimen (above), further converted into (IV; R = OPh, R' = NH₂), m. p. 214—215°, alone or in admixture with an authentic specimen.

The authors thank the Therapeutic Research Corporation of Great Britain Limited for grants and for certain facilities.

QUEEN MARY COLLEGE (UNIVERSITY OF LONDON), E.1.

[Received, December 31st, 1946.]